Johnson, C. K. (1971). ORTEP. Report ORNL-3794, revised Oak Ridge National Laboratory, Tennessee, USA.
Jones, W. \& Theocharis, C. R. (1984). J. Cryst. Spectrosc. Res. 14, 447-455.
Kashino, S., Oka, H. \& Haisa, M. (1989). Acta Cryst. C45, 154-157.
Main, P., Hull, S. E., Lessinger, L., Germain, G., Declerce, J.-P. \& Woolfson, M. M. (1978). MULTAN78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.

Nakanishi, H., Hasegawa, M. \& Mori, T. (1985). Acta Cryst. C41, 70-71
Nakanish, H., Jones, W., Thomas, J. M., Hursthouse, M. B. \& Motevalli, M. (1981). J. Phys. Chem. 85, 3636-3642.
Raghunathan, S. R. \& Pattabhi, V. (1979). Acta Cryst. B35, 214-215.
Raghunathan, S. R. \& Pattabhi, V. (1981). Acta Cryst. B37, 1299-1301.
Sakurai, T. (1967). RSSFR-5. The Universal Crystallographic Computing System (I). Tokyo: The Crystallographic Society of Japan.

Acta Cryst. (1989). C45, 1758-1760

# 2-Phenylsparteine 

By A. Katrusiak, E. Figas, Z. Kaluski and D. Lesiewicz<br>Department of Crystal Chemistry, Faculty of Chemistry, Adam Mickiewicz University, ul. Grunwaldzka 6, 60-780 Poznañ, Poland

(Received 19 January 1989; accepted 8 March 1989)


#### Abstract

C}_{21} \mathrm{H}_{30} \mathrm{~N}_{2}, \quad M_{r}=310.48\), orthorhombic, $P 2_{1} 2_{1} 2_{1}, \quad a=7.586(5), \quad b=14.415$ (2),$\quad c=$ 16.936 (4) $\AA, \quad V=1851.9(4) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.12 \mathrm{~g} \mathrm{~cm}^{-3}, \quad$ m.p. $=352-353 \mathrm{~K}, \quad \lambda($ Mo $K \alpha)=$ $0.71069 \AA, \mu($ Mo $K \alpha)=0.6 \mathrm{~cm}^{-1}, F(000)=680, T=$ 292 K , final $R=0.044$ for 1489 observed reflections. The quinolizidine moieties both have trans configurations, and piperidine rings $A, B, C$ and $D$ have chair, chair, boat and chair conformations, respectively. The phenyl subtituent at $\mathrm{C}(2)$ is equatorial.


Introduction. This structure determination of 2 phenylsparteine (I) continues our studies on sparteine derivatives (Kałuski, Skolik \& Wiewiórowski, 1978; Szymczak, 1983). Our previous X-ray determinations of 2-phenyl derivatives of sparteine included 2-phenylsparteine $N(16)$-oxide monoperchlorate (Małuszyńska \& Okaya, 1977), 2-(p-tolyl)-2-dehydrosparteine (Małuszyńska, Boczoń \& Kałuski, 1986) and $\Delta^{1(2)}$-dehydro-2-phenylsparteine diperchlorate (Boczoń, Kałuski \& Małuszyńska, 1987). The synthesis of 2-phenylsparteine has been described and the configuration of $\mathrm{C}(2)$, as well as the configuration of the flexible $C / D$ moiety, have been determined on the basis of a comparison of IR spectra of this compound with those of monodeuterated derivatives (Boczoń, 1981). The sparteine molecule consists of two quinolizidine moieties (rings $A-B$ and $C-D)$. The configuration of the quinolizidine moieties is trans-trans for $\alpha$-isosparteine, trans-cis for sparteine and cis-cis for $\beta$-isosparteine (for molecules with all four piperidine rings in chair conformation). The trans form is rigid, while the cis form is flexible. Thus, in sparteine and $\beta$-isosparteine the inversion of $\mathrm{N}(16)$, the conversion of ring $C$ from

0108-2701/89/111758-03\$03.00
chair to boat, and the change of configuration of moiety $C / D$ from cis to trans are possible. These compounds, as free bases in the liquid state and in solution, exist predominantly with ring $C$ in the boat form (Bohlmann, Schumann \& Arndt, 1965; Skolik, Krueger \& Wiewiórowski, 1970). The main aim of this X-ray determination was to confirm the spectroscopic results concerning the conformation of ring $C$ and to determine the orientation of the phenyl substituent at $\mathrm{C}(2)$.

Experimental. Crystals of (I) suitable for X-ray analysis were obtained from ethanol solution. Colourless elongated plates, crystal dimensions $0.3 \times 0.3 \times 0.2 \mathrm{~mm}$. CAD-4 diffractometer, graphitemonochromated Mo $K \alpha$ radiation, unit-cell parameters from least-squares fit to 25 reflections $(9 \cdot 9 \leq 2 \theta$ $\leq 13.7^{\circ}$ ), $2 \theta-\theta$ scan method, variable scan speed ( $2 \cdot 0-20 \cdot 0^{\circ} \mathrm{min}^{-1}$ ) depending on reflection intensity. 1895 reflections up to $2 \theta=50^{\circ}$ were measured: $h 0 \rightarrow 9, k 0 \rightarrow 17, l 0 \rightarrow 6$; no systematic variation in intensity was observed for three control reflections measured every $2 \mathrm{~h}, \mathrm{Lp}$ corrections, no absorption corrections, 1489 reflections with $I \geq 1.96 \sigma(I)$ were considered observed. The structure was solved by direct methods with SHELX76 (Sheldrick, 1976) and the absolute configuration of the molecule was assigned according to the previous determination of the absolute configuration of the naturally occurring sparteine derivatives as $\mathrm{C}(7) S$ and $\mathrm{C}(9) S$ (Klyne, Scopes, Thomas, Skolik, Gawroński \& Wiewiórowski, 1974). The positions of all H atoms were recalculated from the geometry of the molecule after each cycle of refinement, all H atoms were assigned an equal isotropic temperature factor of
© 1989 International Union of Crystallography

Table 1. Atomic coordinates and equivalent isotropic thermal parameters ( $\AA^{2}$ )

| $U_{\text {eq }}=\left(U_{11}+U_{22}+U_{33}\right) / 3$ |  |  |  |  |
| :--- | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
|  | $x$ | $0.5343(2)$ | $0.1698(1)$ | 0.044 |
| $\mathrm{~N}(1)$ | $0.1799(3)$ | $0.4346(2)$ | $0.1482(2)$ | 0.047 |
| $\mathrm{C}(2)$ | $0.1639(4)$ | $0.4138(2)$ | $0.1126(2)$ | 0.058 |
| $\mathrm{C}(3)$ | $-0.0169(4)$ | $0.4424(2)$ | $0.1676(2)$ | 0.066 |
| $\mathrm{C}(4)$ | $-0.1674(4)$ | $0.5452(2)$ | $0.1870(2)$ | 0.067 |
| $\mathrm{C}(5)$ | $-0.1448(4)$ | $0.5635(2)$ | $0.2230(2)$ | 0.055 |
| $\mathrm{C}(6)$ | $0.0364(4)$ | $0.6668(2)$ | $0.2436(2)$ | 0.059 |
| $\mathrm{C}(7)$ | $0.0610(4)$ | $0.6793(2)$ | $0.2885(2)$ | 0.060 |
| $\mathrm{C}(8)$ | $0.2332(4)$ | $0.6518(2)$ | $0.2317(2)$ | 0.049 |
| $\mathrm{C}(9)$ | $0.3786(4)$ | $0.5506(2)$ | $0.2071(2)$ | 0.051 |
| $\mathrm{C}(10)$ | $0.3520(4)$ | $0.7194(2)$ | $0.1606(2)$ | 0.054 |
| $\mathrm{C}(11)$ | $0.3813(4)$ | $0.7802(2)$ | $0.1627(2)$ | 0.075 |
| $\mathrm{C}(12)$ | $0.5462(5)$ | $0.8497(3)$ | $0.0944(3)$ | 0.097 |
| $\mathrm{C}(13)$ | $0.5483(6)$ | $0.9058(3)$ | $0.0940(3)$ | 0.101 |
| $\mathrm{C}(14)$ | $0.3810(7)$ | $0.8425(3)$ | $0.0943(3)$ | 0.095 |
| $\mathrm{C}(15)$ | $0.2218(6)$ | $0.7808(2)$ | $0.1624(2)$ | 0.060 |
| $\mathrm{~N}(16)$ | $0.2278(4)$ | $0.7293(2)$ | $0.1696(2)$ | 0.067 |
| $\mathrm{C}(17)$ | $0.0617(4)$ | $0.4065(2)$ | $0.0902(2)$ | 0.048 |
| $\mathrm{C}(18)$ | $0.3059(4)$ | $0.3218(2)$ | $0.0980(2)$ | 0.060 |
| $\mathrm{C}(19)$ | $0.3920(5)$ | $0.2938(3)$ | $0.0427(2)$ | 0.076 |
| $\mathrm{C}(20)$ | $0.5165(5)$ | $0.3496(3)$ | $-0.0200(3)$ | 0.084 |
| $\mathrm{C}(21)$ | $0.5566(5)$ | $0.4337(3)$ | $-0.0288(2)$ | 0.079 |
| $\mathrm{C}(22)$ | $0.4725(6)$ | $0.4619(2)$ | $0.0256(2)$ | 0.065 |
| $\mathrm{C}(23)$ | $0.3479(5)$ | 0.4610 |  |  |

$0.07 \AA^{2}$. The function minimized was $\sum w\left(\left|F_{o}\right|-\right.$ $\left.\left|F_{c}\right|\right)^{2}, w=\sigma^{-2}\left(F_{o}\right)$, final $R=0.044, w R=0.039, S=$ $1 \cdot 7,(\Delta / \sigma)_{\max }=0.002$, the highest and the lowest peaks on the final $\Delta F$ map were 0.11 and $-0.14 \mathrm{e} \AA^{-3}$. Most of the calculations were performed with SHELX76 (Sheldrick, 1976), atomic scattering parameters as incorporated in this program, IBM XT-Turbo computer. The final atomic parameters are listed in Table 1.*

Discussion. The bond lengths and valency angles are given in Fig. 1. They are similar to those observed in other sparteine derivatives; charactertistic features are sharpening of angle $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ and widening of $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(17)$ and $\mathrm{C}(10)-$ $\mathrm{C}(9)-\mathrm{C}(11)$ or $\mathrm{N}(1)-\mathrm{C}(10)-\mathrm{C}(9)$ and $\mathrm{C}(7)-\mathrm{C}(17)$ $-\mathrm{N}(16)$ in the region of fusion of the two quinollizidine moieties. There are no significant differences in bond lengths and valency angles between the two quinolizidine moieties, except for bonds about $\mathrm{C}(2)$ and $\mathrm{C}(15)$. The $\mathrm{C}(2)-\mathrm{C}(18)$ bond length agrees with the length of a single $\mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{2}}$ bond ( $1.510 \AA$; Allmann, 1977). In the phenyl ring the largest departure from the ideal angle of $120^{\circ}$ is observed for $\mathrm{C}(19)-\mathrm{C}(18)-\mathrm{C}(23)$, as was also observed for 2-( $p-$ tolyl)-2-dehydrosparteine (Małuszyńska, Boczoñ \& Kałuski, 1986). No significant distortion from planarity is observed for the phenyl ring.

[^0]

Fig. 1. Bond lengths $(\AA \AA)$ and valency angles $\left({ }^{\circ}\right)$.


Fig. 2. Torsion angles $\left({ }^{\circ}\right)$. The endocyclic torsion angles are given inside the rings while the exocyclic angles are given outside the molecular contour and are defined by the bonds along the outline of the molecule.



Fig. 3. A stereodiagram of the unit-cell contents along a (PLUTO; Motherwell, 1976). H atoms are omitted for clarity.

The torsion angles are given in Fig. 2. The piperidine rings $A, B, C$ and $D$ have chair, chair, boat and chair conformations, respectively. The position of the phenyl ring can be conveniently described by the torsion angles along $\mathrm{C}(2)-\mathrm{C}(18)$ (see Fig. 2). The corresponding torsion angles in 2-phenylsparteine $N(16)$-oxide perchlorate (Małuszyńska \& Okaya, 1977) have very similar valules of 137.0 (6) and $76.8(8)^{\circ}$; a similar orientation of the phenyl ring was also observed in 2 -( $p$-tolyl)-2-dehydrosparteine [torsion angles $132 \cdot 9(7)$ and $-44 \cdot 6(9)^{\circ}$ ].

A stereoview of the unit-cell contents of (I) is presented in Fig. 3. No intermolecular contacts shorter than van der Waals distances were observed.

We are grateful to Professor M. Wiewiórowski for suggesting this investigation, to Dr Wł. Boczoń for supplying the compound and to Dr M. Wieczorek, of the Technical University, Łódź, for collecting the diffractometer data. This study was supported by the Polish Ministry of Science and Higher Education, Project RP II-10.

## References

Allmann, R. (1977). In Homoatomic Rings, Chains and Macromolecules of Main Group Elements, edited by A. Rheingold. Amsterdam: Elsevier.
Boczoń, We. (1981). Pol. J. Chem. 55, 339-351.
Boczoń, We., Kaluski, Z. \& Maeuszyńska, H. (1987). Bull. Acad. Pol. Sci. Sér. Sci. Chim. 35, 541-550.
Bohlmann, F., Schumann, D. \& Arndt, C. (1965). Tetrahedron Lett. pp. 2705-2711.
Kaluski, Z., Skolik, J. \& Wiewiórowski, M. (1978). Proceedings, Precongress Symposium on Organic Crystal Chemistry, Dymaczewo, Poland, pp. 321-343.
Klyne, W., Scopes, P. N., Thomas, R. N., Skolik, J., Gawroński, J. \& Wiewiórowski, M. (1974). J. Chem. Soc. Perkin Trans. 1, pp. 2565-2570.
Maluszińska, H., Boczoń, We. \& Kaluski, Z. (1986). J. Cryst. Spectrosc. Res. 16, 687-694.
Maluszyńska, H. \& Okaya, Y. (1977). Acta Cryst. B33, 3889 3891.

Motherwell, W. D. S. (1976). PLUTO. Program for plotting molecular and crystal structures. Univ. of Cambridge, England.
Sheldrick, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England.
Skolik, J., Krueger, P. J. \& Wiewiórowski, M. (1970). J. Mol. Struct. 5, 461-475.
Szymczazk, A. (1983). Thesis, Adam Mickiewicz Univ., Poznań, Poland.

# Non-Natural 14-Hydroxy Steroids. I. Methyl 14 $\beta$-Hydroxy-1,7,17-trioxo-5 $\beta, 8 \alpha, 9 \beta$-androstan-10 $\beta$-oate 

By André G. Michel,* $\dagger$ Résean Ruel $\ddagger$ and Nadine Michel-Dewez $\dagger$<br>Laboratoire de chimie structurale and Laboratoire de chimie organique, Université de Sherbrooke, Sherbrooke, Québec Canada J1K $2 R 1$

(Received 28 December 1988; accepted 15 March 1989)


#### Abstract

C}_{20} \mathrm{H}_{26} \mathrm{O}_{6}, M_{r}=362 \cdot 42\), monoclinic, $A 2 / n$, $a=16.6256$ (8), $\quad b=10.6293$ (5),$\quad c=20.3059$ (6) $\AA$, $\beta=91.515(3)^{\circ}, \quad V=3587.17$ (7) $\AA^{3}, \quad D_{x}=$ $1.342 \mathrm{Mg} \mathrm{m}^{-3}, \quad Z=8, \quad \lambda(\mathrm{Cu} K \bar{\alpha})=1.54056 \AA, \quad \mu=$ $0.77 \mathrm{~mm}^{-1}, F(000)=1552$, room temperature, final $R=0.039$ for 3207 observed reflections. The nonnatural steroid compound bears a methoxycarbonyl group at $\mathrm{C}(10)$. The relative stereochemistry is cis for the $A / B$ ring junction, syn between $\mathrm{MeO}_{2} \mathrm{C}-\mathrm{C}(10)$ and $\mathrm{H}-\mathrm{C}(9)$, trans for the $B / C$ ring junction, anti between $\mathrm{H}-\mathrm{C}(8)$ and $\mathrm{HO}-\mathrm{C}(14)$ and cis for the $C / D$ ring junction, and an all-chair conformation is adopted.


Introduction As part of a study aimed at the synthesis of various natural and non-natural 14-hydroxy

[^1]steroids, compound (1) was obtained upon acidic or alkaline treatment of the tetraketone (2) (Ruel \& Deslongchamps, 1988). One could predict the stereochemical identity of the $\mathrm{C}(14)$ carbon center as that shown in structure (1), based on related work (Yates, Douglas, Datta \& Sawyer, 1988) which reported the synthesis of the $14 \beta$-hydroxy steriod (3) by a similar approach. Unequivocal assignment of the structural identity of the steroidal compound (3) was made by X-ray analysis (Douglas, Sawyer \& Yates, 1987). The present crystallographic analysis was undertaken to confirm the predicted structure of steriod (1) (Fig. 1).

Experimental. Crystal $0.20 \times 0.15 \times 0.30 \mathrm{~mm}$; EnrafNonius CAD-4 diffractometer graphite-monochromator, $\mathrm{Cu} K \bar{\alpha}$ radiation; cell dimensions were obtained from 28 reflections with $2 \theta$ angles in the range $80 \cdot 0-120 \cdot 0^{\circ}$. The $\omega / 2 \theta$ scan mode was used for data collection at a constant scan speed of $4^{\circ} \mathrm{min}^{-1}$.
(c) 1989 International Union of Crystallography


[^0]:    * Lists of structure factors, anisotropic thermal parameters and H -atom positions have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51996 (12 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

[^1]:    *To whom correspondence should be addressed.
    $\dagger$ Laboratoire de chimie structurale.
    $\ddagger$ Laboratoire de chimie organique.

